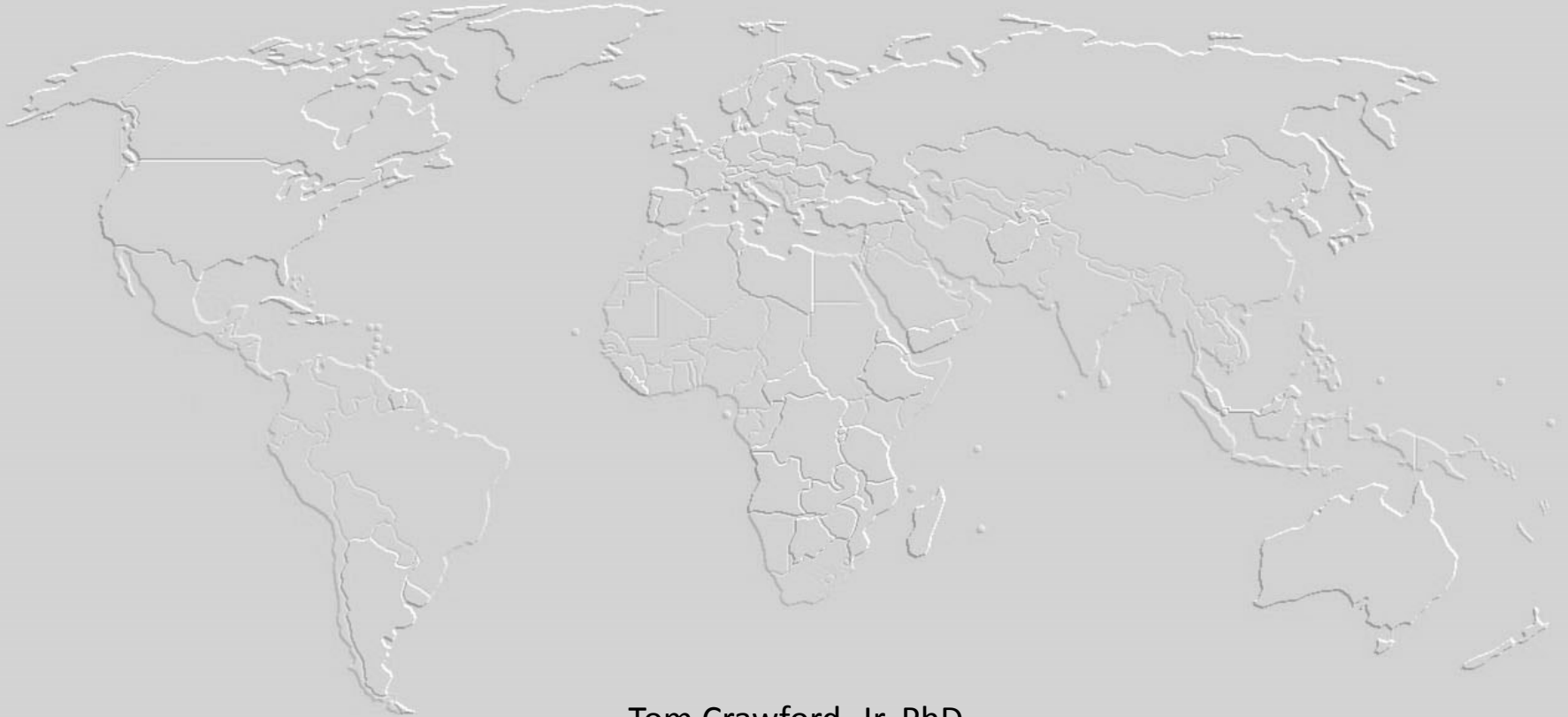


How Irrigation Water Quality and Crop Nutrition Impact Food Security in Developing Countries



Tom Crawford, Jr, PhD

Bio Huma Netics, Inc.

Washington, DC, October 23, 2013

Food Security

USAID Policies

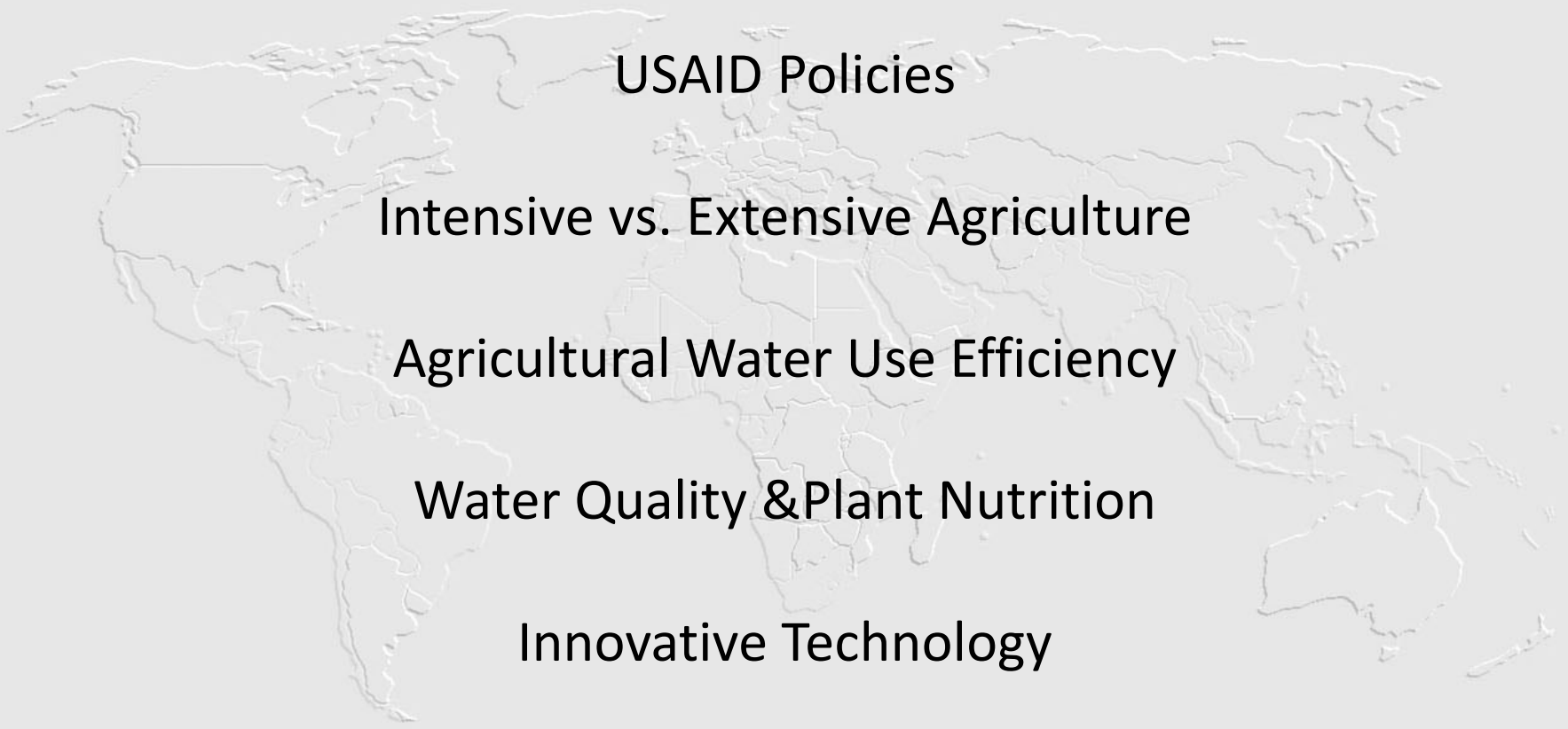
Intensive vs. Extensive Agriculture

Agricultural Water Use Efficiency

Water Quality & Plant Nutrition

Innovative Technology

Sustainable & Profitable Value Chains



What is Food Security?

Food security is considered to exist when the following four components (pillars) have been attained:

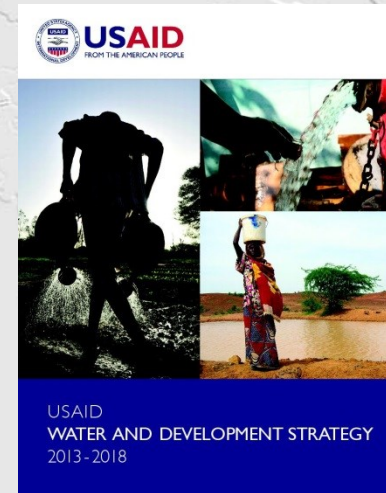
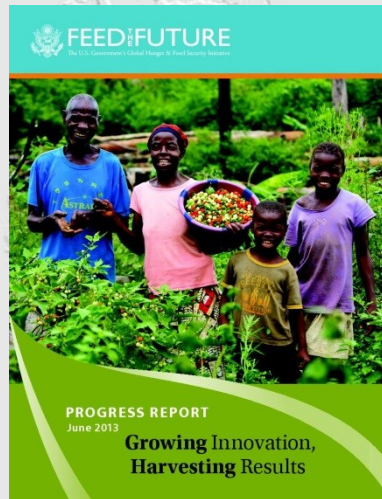
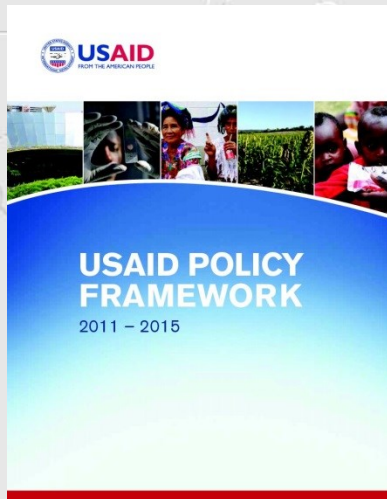
- **Availability:** Families and individuals require a reliable and consistent source of quality food.
- **Access:** Sufficient resources (purchasing power) to purchase quality foods.
- **Utilization:** People need to have knowledge and basic sanitary conditions in order to choose, prepare and distribute the quality foods in a manner that results in good nutrition for all family members.
- **Stability:** A stable and sustainable environment (social, political and economy) must exist in order to ensure that families and individuals have the availability, access and utilization of quality and nutritious foods.

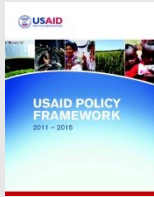
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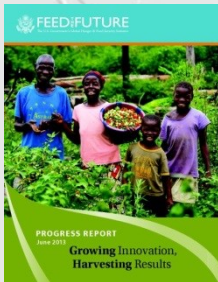
USAID Policies Support Food Security and Improved Management of Water





USAID Policy Framework 2011-2015

“APPLY SCIENCE, TECHNOLOGY, AND INNOVATION STRATEGICALLY.”



“Scientific innovation and technology are critical to meeting the global challenges of producing more food with less **land** and **water** and helping farmers adapt to climatic, social and economic shocks.”

USG's Feed the Future Policy Supports Application of Scientific Innovation and Technology for Intensification of Agriculture

Sub-Saharan Africa – 40 years of *extensive* cereal production

Asia – 40 years of *intensive* cereal production

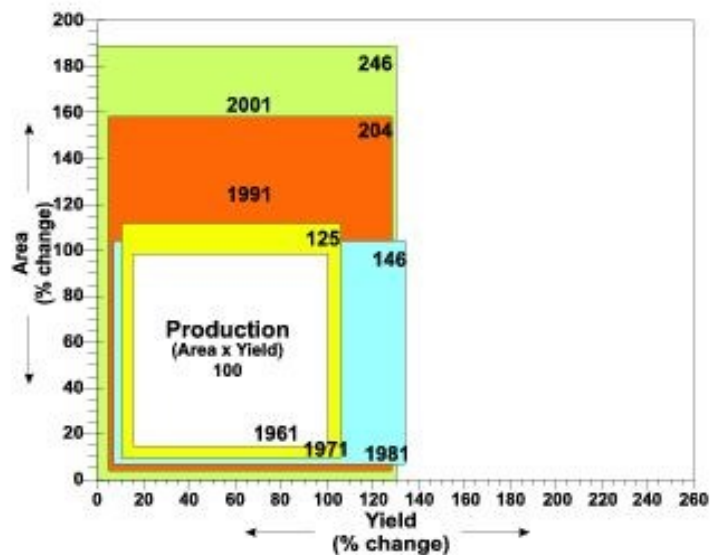


Figure 2. Changes in Cereal Production in Sub-Saharan Africa Due to Changes in Area and Yield (1961 = 100)

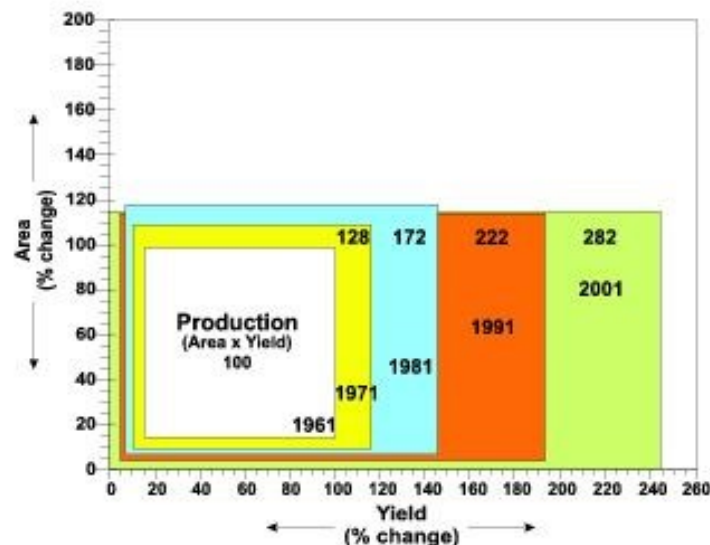
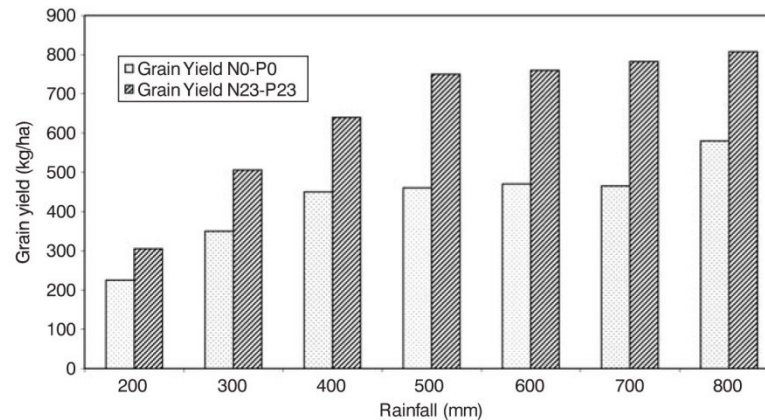


Figure 3. Change in Cereal Production in Asia Due to Changes in Area and Yield (1961 = 100)

Intensification of Pearl Millet Production in Niger Using Irrigation & Fertilizer – No Increase in Area Cultivated

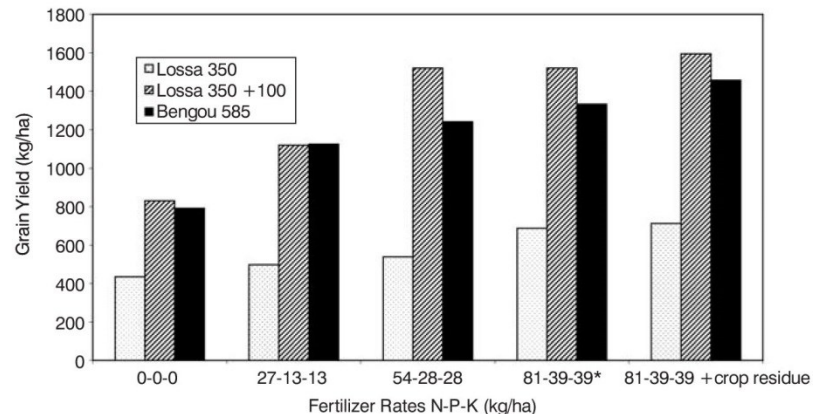
Yield of rainfed pearl millet increased with more rainfall and with N & P fertilization.

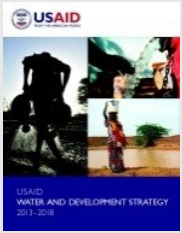
FIGURE 6. Pearl millet grain yield response to nitrogen and phosphorus in different rainfall zones in Niger as an average of more than 500 site years.



Yield of rainfed pearl millet increased with either greater rainfall or supplementary irrigation and fertilization.

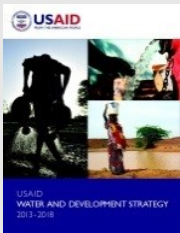
FIGURE 7. Pearl millet grain yield at two rainfall zones in Niger, and effect of supplementary irrigation in the low rainfall zone in 1998.





USAID Water and Development Strategy 2013 - 2018

- Support for Feed the Future
- Strategic Objectives 2013 – 2018
 - SO 1 – Improve health outcomes through the provision of sustainable safe water, sanitation, and hygiene (WASH)
 - SO 2 - Manage water in agriculture sustainably and more productively to enhance food security
 - Intermediate Result 2.1: Improve the efficiency and sustainability of food production from rainfed agricultural systems.
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Water Use Efficiency, or Productivity, California, 1989 – 2009

WUE = tons/acre-foot applied (35.3% increase, or ~1.76% average increase per annum)

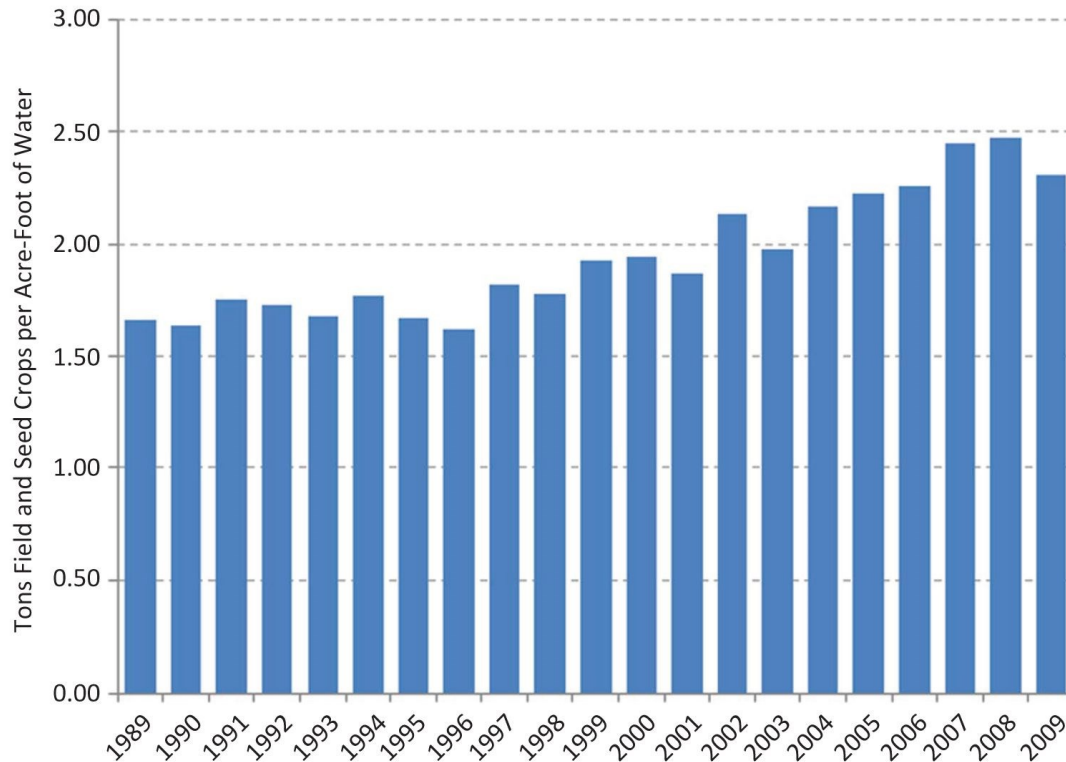
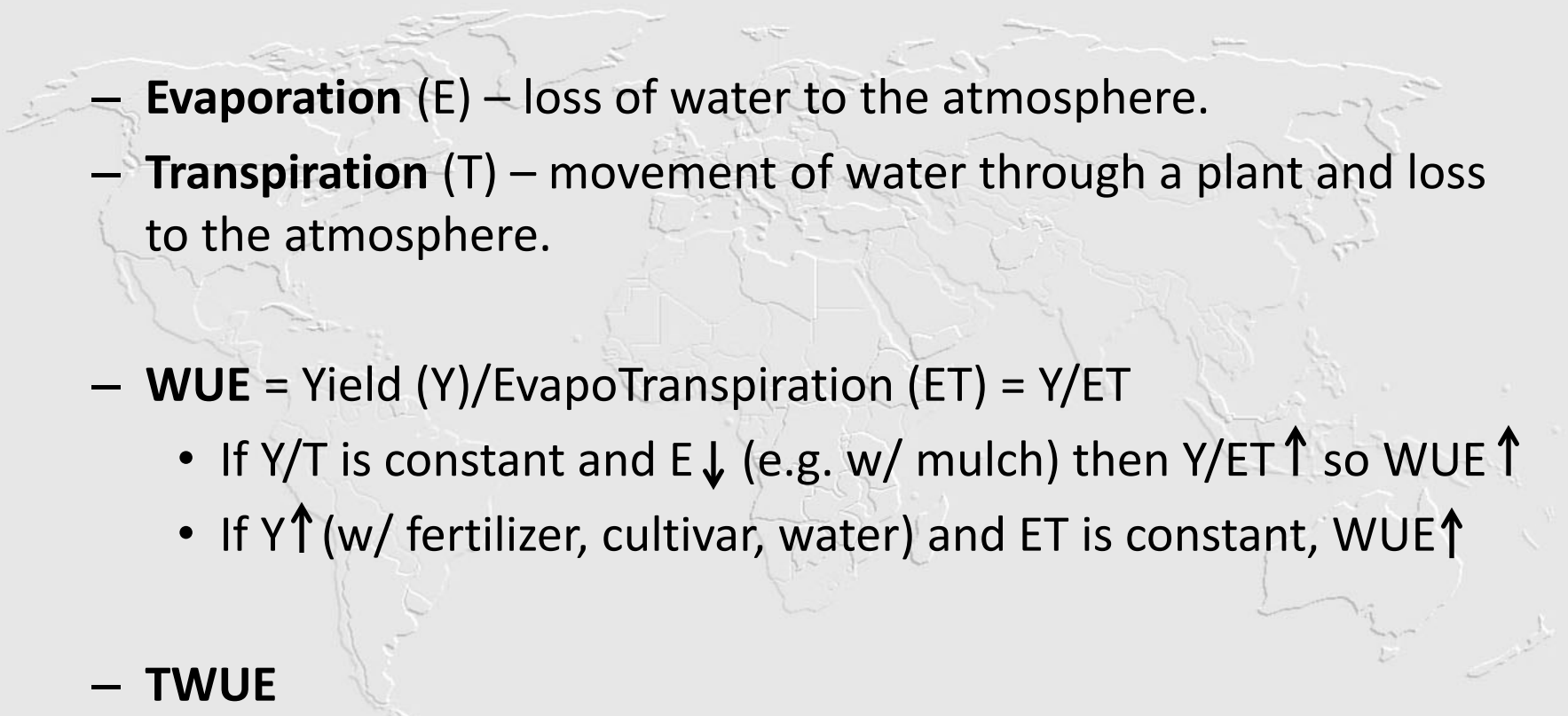


Figure 2. California crop productivity for field/seed crops (tons per acre-foot of water). Productivity has increased from around 1.7 to over 2.3 or more between the late 1980s and 2009 without a comparable increase in total water use for these crops.

Sources: USDA NASS CA Historical Data (1989–2008) and CA Agricultural Statistics Report (2009).

Water Use Efficiency (WUE) and Transpirational Water Use Efficiency (TWUE)

- 
- **Evaporation** (E) – loss of water to the atmosphere.
 - **Transpiration** (T) – movement of water through a plant and loss to the atmosphere.
 - **WUE** = Yield (Y)/EvapoTranspiration (ET) = Y/ET
 - If Y/T is constant and $E \downarrow$ (e.g. w/ mulch) then $Y/ET \uparrow$ so $WUE \uparrow$
 - If $Y \uparrow$ (w/ fertilizer, cultivar, water) and ET is constant, $WUE \uparrow$
 - **TWUE**
 - Transpirational WUE, $TWUE = Y/T \uparrow$
 - If $Y/T \uparrow$ (fertilizer, cultivar), $TWUE \uparrow$

Constraints to Agricultural WUE

- Social, political and economic
 - Laws, regulations, cultural practices, beliefs.
 - Lack of access to inputs & facilities due to lack of investment, infrastructure and functioning value chains.
- Physical and chemical
 - Biotic stresses (e.g. pests, pathogens)
 - Abiotic stresses (e.g. deficiencies or excesses in moisture, temperature, water quality, essential plant nutrients, and other elements in the environment)

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**Effects of Irrigation Water Quality
on Water Use Efficiency**

Irrigation Water Quality Characteristics

1. Concentration of soluble salts.*
2. Concentration of sodium and proportion of sodium to calcium plus magnesium.*
3. Concentration of bicarbonate.
4. Occurrence of minor elements, such as boron, in amounts that are toxic.

*The two water quality factors which have the greatest influence on water infiltration rate.

Salinity- the Concentration of Soluble Salts

- **Salts in irrigation water or soil reduce water availability to such an extent that yield is diminished.**
- Salt, or salinity, in irrigation water is composed of cations and anions so that the sum of the positive (+) and negative (-) charges is electrically neutral.
- *Total concentration of salts = sum of cations and anions.*
- *Important cations and anions in irrigation water:*
 - Cations: H^+ , Na^+ , K^+ , Ca^{2+} , Mg^{2+}
 - Anions: OH^- , HCO_3^- , CO_3^{2-} , SO_4^{2-} , Cl^- , NO_3^- , HBO_3^{2-}
- **Measurement of salinity**
 - Weight per unit volume (ppm, lbs/acre-foot).
 - Electrical equivalents per unit volume (meq/L, μ eq/ml)
 - Electrical Conductivity (mmhos/cm, μ mhos/cm, dS/m).
 - Osmotic Potential (bar, atm, kPa).

Sodium, Calcium & Magnesium – Important to Soil Structure, Health and Crop Yield

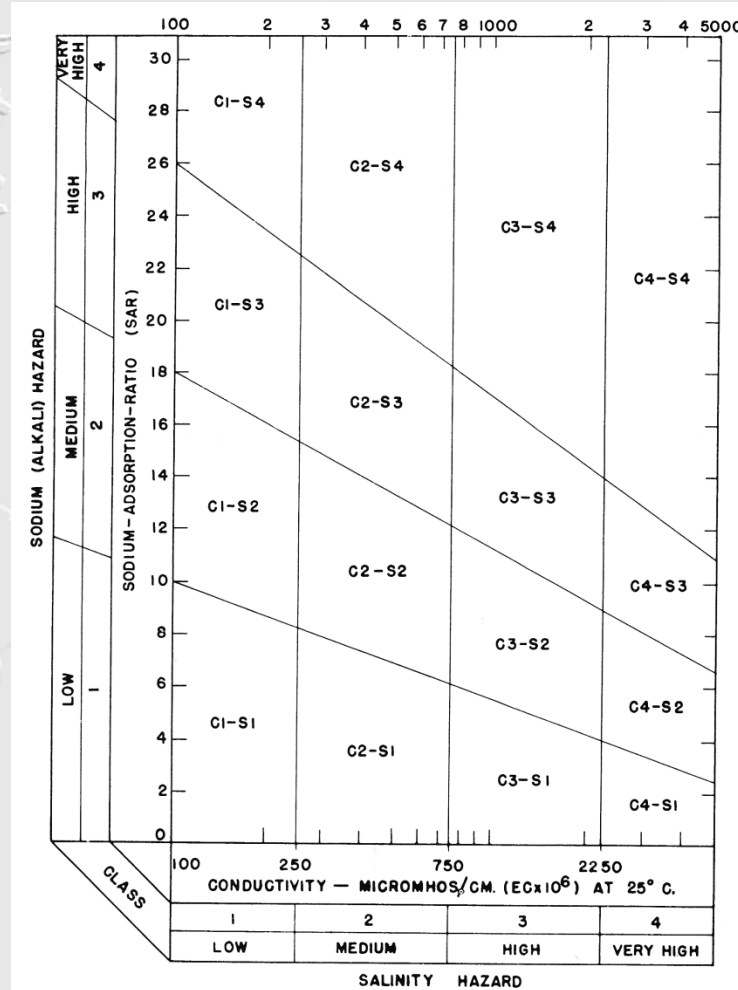
- **Sodium tends to disperse colloidal particles (<0.002 mm) in a soil, resulting in:**
 - impaired gas exchange (particularly O₂ and CO₂).
 - reduced infiltration rates of water and slower water movement in all directions in the soil.
 - Reduced rates of uptake of water, mineral nutrients and oxygen by plants.
 - Diminished rates of crop growth and diminished crop yield.
- **Calcium and magnesium tend to aggregate colloidal particles in a soil, resulting in:**
 - Creation of pore space.
 - Freer movement of gases in the soil.
 - Greater oxygen content of soil pores.
 - Increased aerobic microbial activity.
 - Increased rates of uptake of oxygen, water and mineral nutrients by plant roots.
 - Increased crop growth rates and yield.

Sodium Adsorption Ratio (SAR)

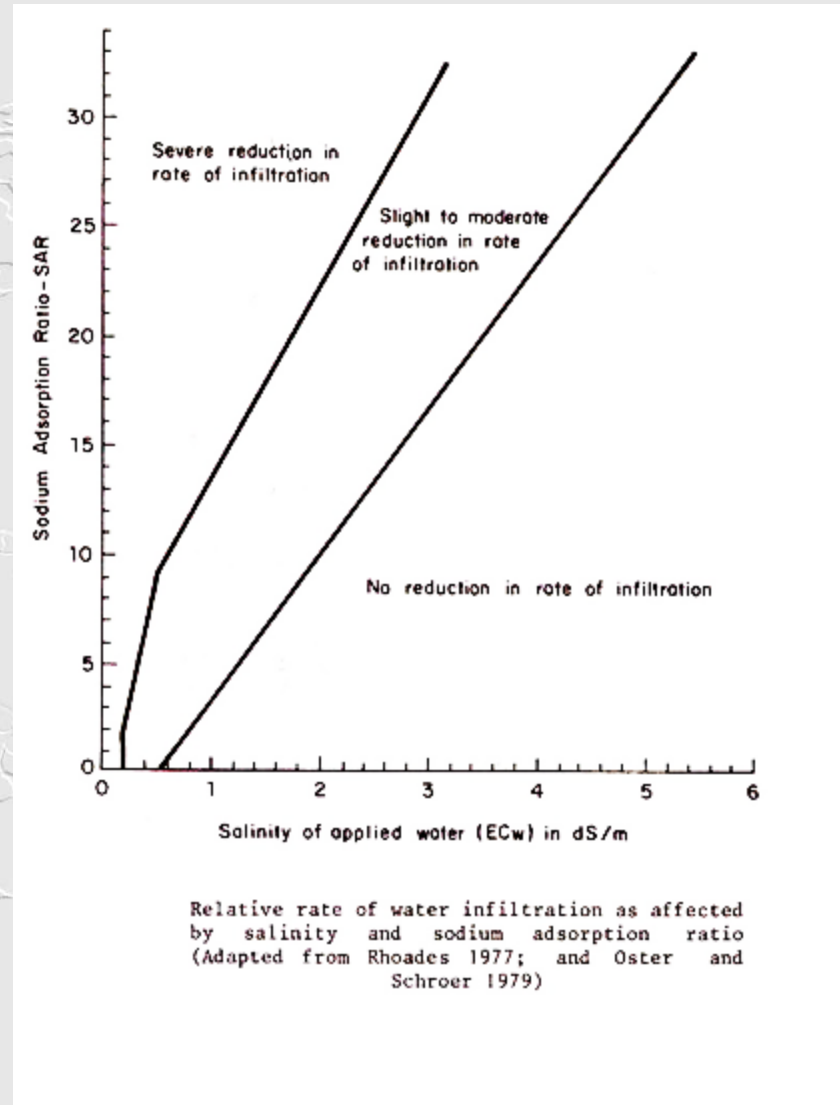
- Excessive Na in irrigation water (>3:1::Na:Ca)
- SAR is the most commonly used method to evaluate the infiltration rate problem.
- $$\text{SAR} = \text{Na}^+ / \sqrt{(\text{Ca}^{2+} + \text{Mg}^{2+})/2}$$

where Na^+ , Ca^{2+} and Mg^{2+} represent the concentrations, in millequivalents per liter, of the respective ions.

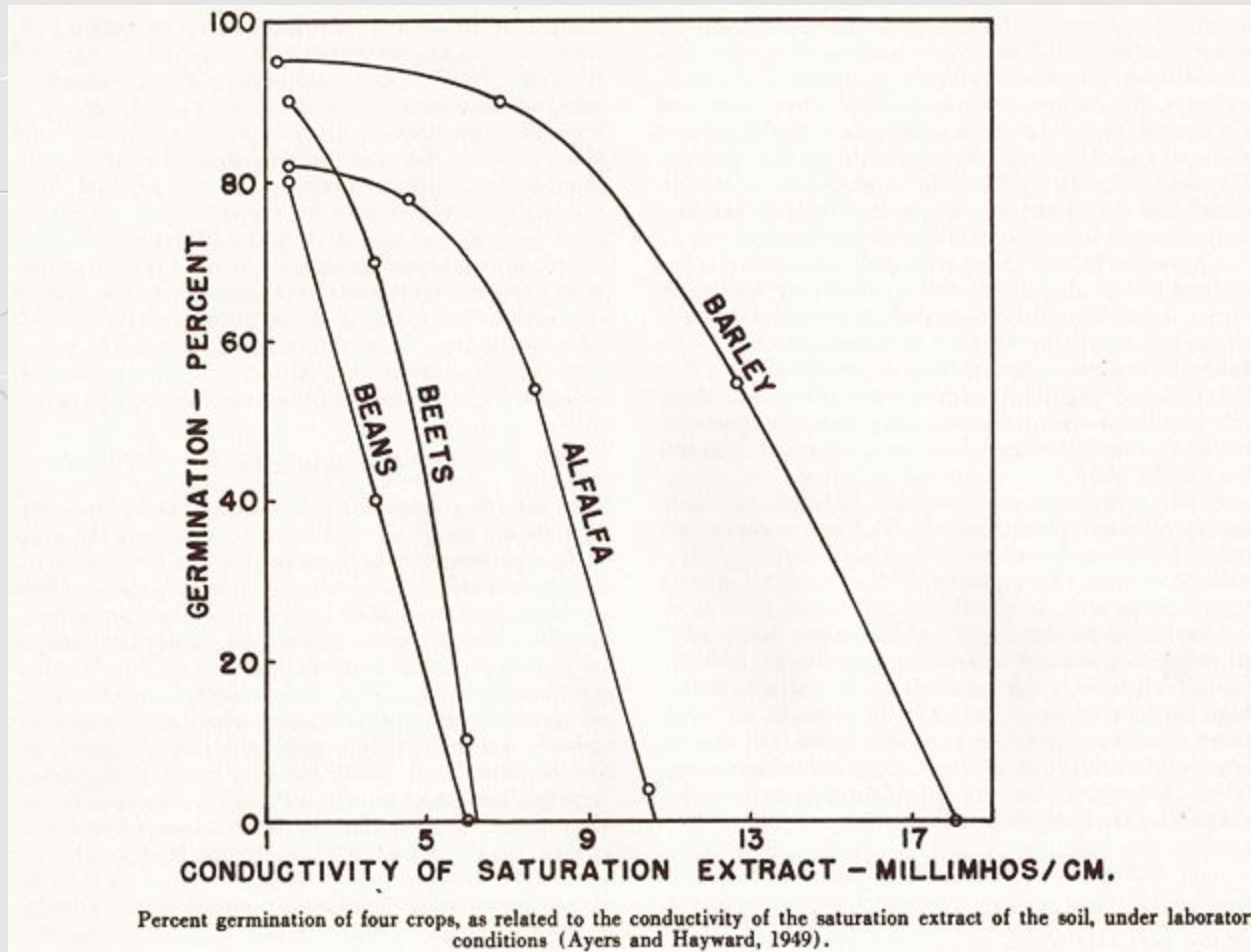
Classification of Irrigation Waters Sodium (Alkali) and Salinity Hazards Based on Sodium Adsorption Ratio and Salinity



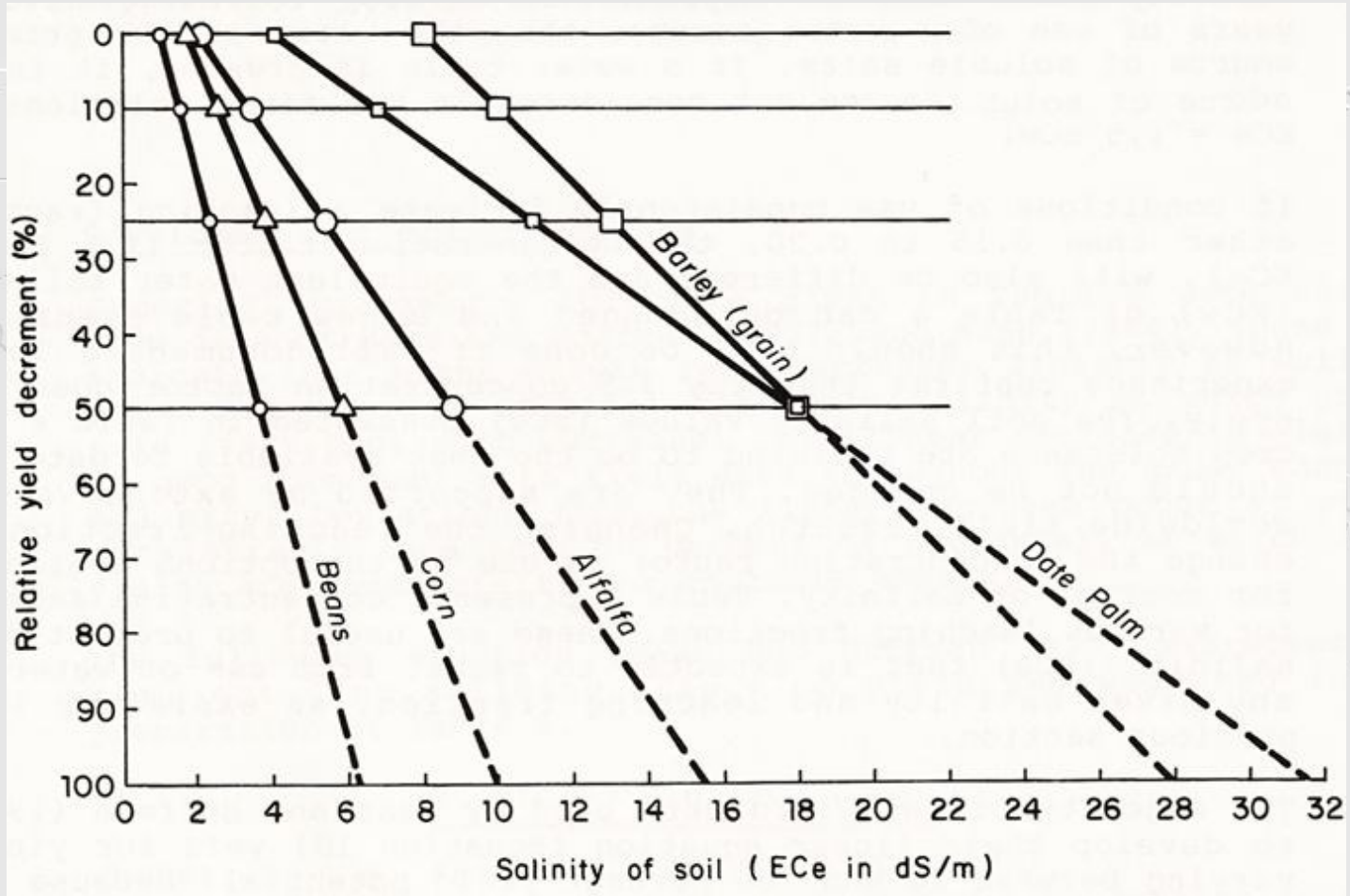
SAR and Salinity to Predict Infiltration Problems



Germination & Soil Salinity



Crop Selection & Soil Salinity



Bicarbonate

- Bicarbonate ions form as a result of the solution of carbon dioxide in water.
- CO₂ may be of atmospheric or biological origin.
- Relative proportions of carbonic acid, bicarbonate ions and carbonate ions depend upon pH of the water.
- **Carbonate from irrigation water or from bicarbonate can form calcium carbonate, or lime, that can clog soil pores.**
- In managing alkaline soils, it is important to reduce or eliminate bicarbonate and carbonate from irrigation water and from the soil solution.

Boron (B), an Essential, but Sometimes Toxic Element

- The principal source of B is the mineral, tourmaline, a widespread but minor constituent of primary rocks.
- Calcium borate can precipitate in alkaline soils, giving rise to boron deficiency.
- Boron exists as borate anions in the soil solution

Relative tolerance of plants to boron

[In each group, the plants first named are considered as being more tolerant and the last named more sensitive]

Tolerant	Semitolerant	Sensitive
Athel (<i>Tamarix aphylla</i>)	Sunflower (native)	Pecan
Asparagus	Potato	Black walnut
Palm (<i>Phoenix canariensis</i>)	Acala cotton	Persian (English) walnut
Date palm (<i>P. dactylifera</i>)	Pima cotton	Jerusalem artichoke
Sugar beet	Tomato	Navy bean
Mangel	Sweetpea	American elm
Garden beet	Radish	Plum
Alfalfa	Field pea	Pear
Gladiolus	Ragged Robin rose	Apple
Broadbean	Olive	Grape (Sultanina and Malaga)
Onion	Barley	Kadota fig
Turnip	Wheat	Persimmon
Cabbage	Corn	Cherry
Lettuce	Milo	Peach
Carrot	Oat	Apricot
	Zinnia	Thornless black-berry
	Pumpkin	Orange
	Bell pepper	Avocado
	Sweetpotato	Grapefruit
	Lima bean	Lemon

Permissible limits of boron for several classes of irrigation waters: <1.0 – >3.5 ppm | <0.67 – >2.5 ppm | <0.33 - > 1.25 ppm



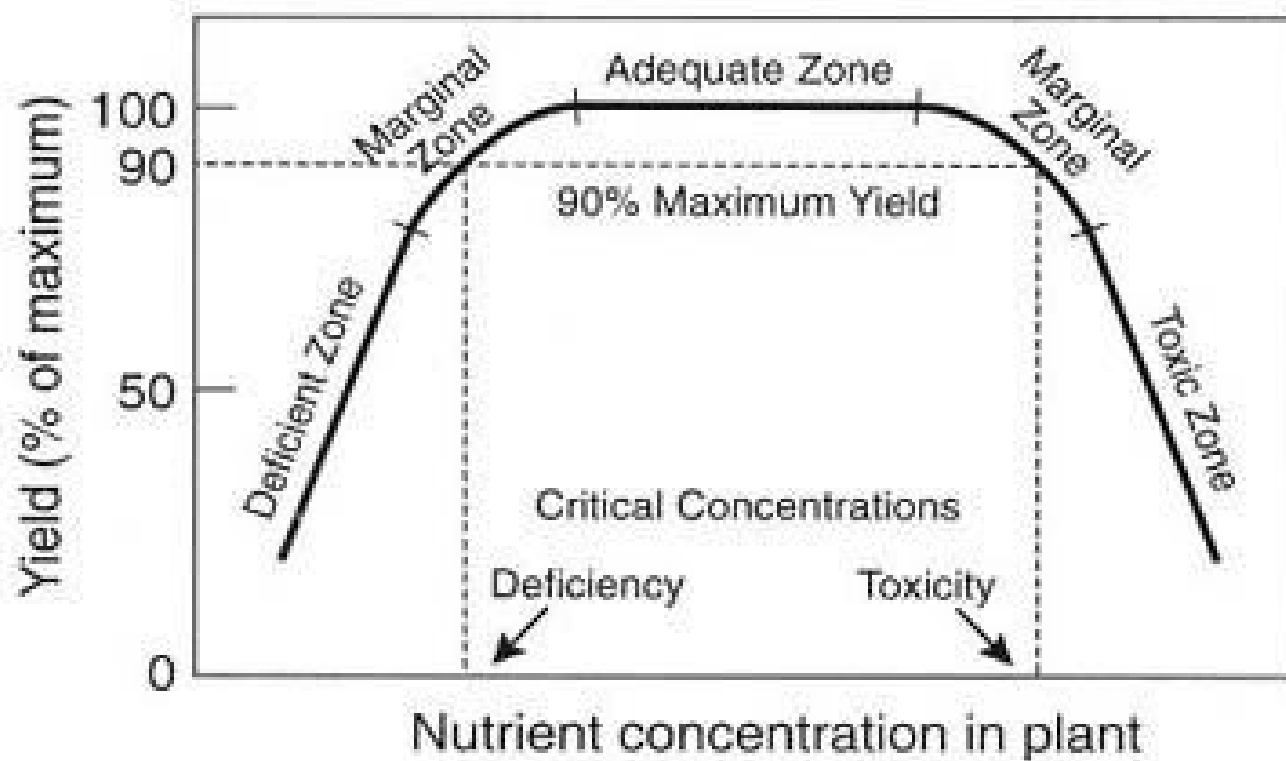
Effects of Crop Nutrition on Water Use Efficiency

The 17 Essential Plant Nutrients in the Periodic Table of the Elements

H																	He							
Li	Be																F	Ne						
Na	Mg																							
K	Ca	Sc	Ti	V	Cr	Mn	Fe								Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe							
Ca	Ba	Lantha nide Metals	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn							
Fr	Ra	Actin ide Metals																						

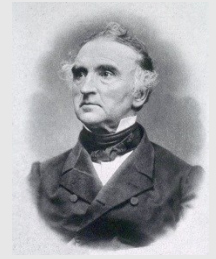
Source: Western Fertilizer Handbook, Seventh Edition

Concentrations of the 17 Essential Plant Nutrients for Maximum Yield



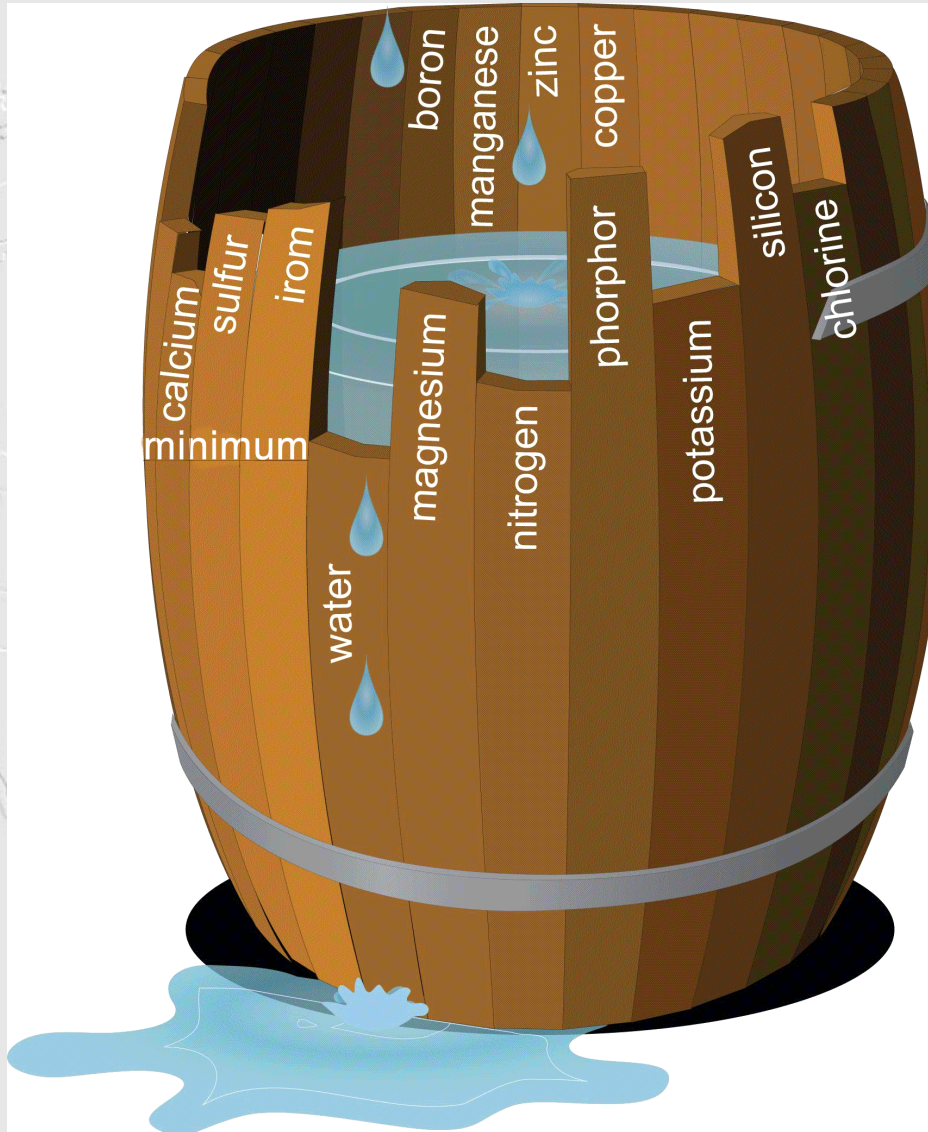


Sprengel-Liebig “Law of the Minimum”



- Theory of Kurt Sprengel (March 29, 1787 – April 19, 1859) popularized by Justus von Liebig (May 12, 1803 – April 18, 1873)
- Sprengel’s “Theory of the minimum”
- *Plant growth is limited by the essential nutrient at the lowest concentration.*
- *Growth is controlled not by the total amount of resources available, but by the scarcest resource, or limiting factor.*

Law of the Minimum



Reduction of Water Use Efficiency Due to Deficiency of Essential Plant Nutrients

Deficiencies of all plant nutrients reduce water use efficiency (WUE), because all nutrient deficiencies reduce the rates of metabolic processes, all of which are in some way connected to photosynthesis that assimilates atmospheric carbon from carbon dioxide and splits water molecules for assimilation of hydrogen and oxygen.

Deficiencies of Essential, Mineral Nutrients in Tomato

An element is essential if it fulfills either one or both of two criteria: (1) The element is part of a molecule that is an intrinsic component of the structure or metabolism of a plant; (2) the plant can be so severely deprived of the element that it exhibits abnormalities in its growth, development, or reproduction—that is, its "performance"—in comparison with plants not so deprived (Epstein and Bloom, 2005).

Epstein, Emanuel and Arnold Bloom. 2005. Mineral nutrition of plants: Principles and perspectives. Sinauer Associates, Inc. Sunderland, Massachusetts

The tomato plants labeled "control" received adequate amounts of 16 essential, mineral nutrients. The tomato plants showing nutrient deficiencies were grown under the same conditions as the control, except for the deficiency of the nutrient (or nutrients in the case of hydrogen and oxygen) indicated. Carbon, an essential element absorbed mainly through the leaves as a component of carbon dioxide, was equally available to all the plants shown. Deficiencies of nickel and molybdenum, the two other essential, mineral nutrients, are not shown.

Photos by T.W. Crawford, Jr.



Hydrogen (H) & Oxygen (O)



Nitrogen (N)



Phosphorus (P)



Potassium (K)



Calcium (Ca)



Magnesium (Mg)



Sulfur (S)



Copper (Cu)



Iron (Fe)



Manganese (Mn)



Zinc (Zn)



Boron (B)



Chlorine (Cl)

Hydrogen & Oxygen (Water) Deficiency

TABLE 3.3 Adequate Concentrations of Elements in Plant Tissue

Element	Chemical symbol	Atomic weight	Concentration in dry matter		Relative number of atoms with respect to nickel
			$\mu\text{mol g}^{-1}$	ppm or %	
Micronutrients					
Nickel	Ni	58.69	0.001	0.05 ppm	1
Molybdenum	Mo	95.95	0.001	0.1 ppm	1
Cobalt	Co	58.94	0.002	0.1 ppm	2
Copper	Cu	63.54	0.10	6 ppm	100
Zinc	Zn	65.38	0.30	20 ppm	300
Sodium	Na	22.91	0.40	10 ppm	400
Manganese	Mn	54.94	1.0	50 ppm	1000
Boron	B	10.82	2.0	20 ppm	2000
Iron	Fe	55.85	2.0	100 ppm	2000
Chlorine	Cl	35.46	3.0	100 ppm	3000
Macronutrients					
Silicon	Si	28.09	30	0.1%	30,000
Sulfur	S	32.07	30	0.1%	30,000
Phosphorus	P	30.98	60	0.2%	60,000
Magnesium	Mg	24.32	80	0.2%	80,000
Calcium	Ca	40.08	125	0.5%	125,000
Potassium	K	39.10	250	1.0%	250,000
Nitrogen	N	14.01	1000	1.5%	1,000,000
→ Oxygen	O	16.00	30,000	45%	← 30,000,000 ←
→ Carbon	C	12.01	40,000	45%	← 40,000,000 ←
→ Hydrogen	H	1.01	60,000	6%	← 60,000,000 ←

Source: Modified from Stout 1961.



Water Deficiency in the Sahel

A climatic deficit of H and O (H₂O) . . .

FIGURE 1. Average monthly potential evapotranspiration (ETP) and rainfall at Konni and Maradi from 1980-1997.

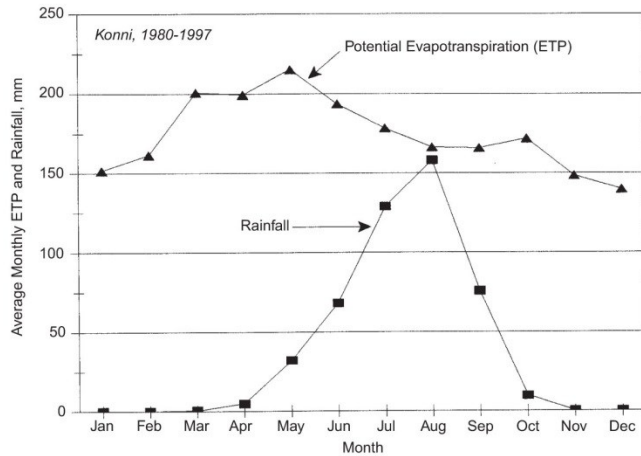
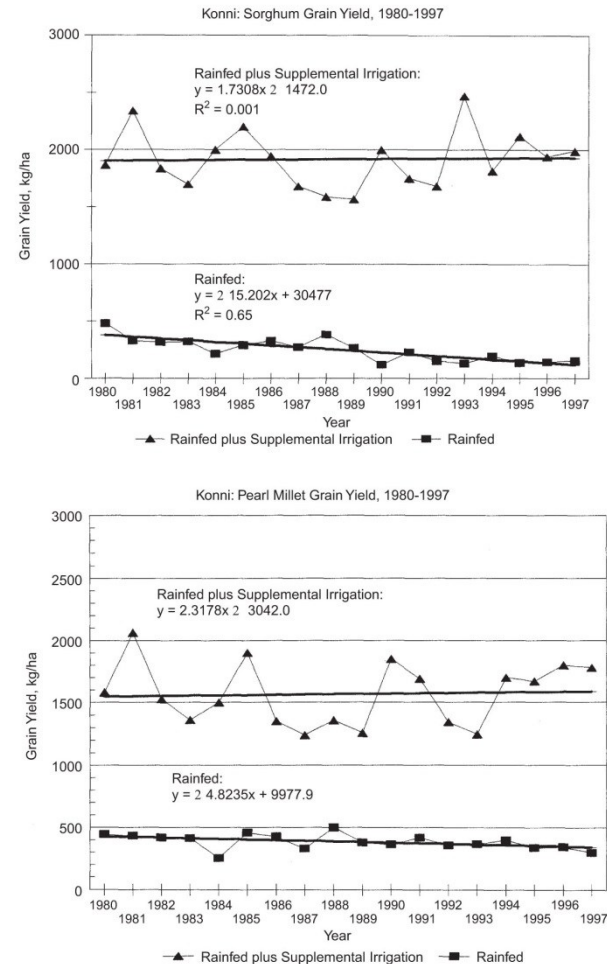


FIGURE 2. Trends in grain yields of pearl millet and sorghum under conditions of rainfall and rainfall plus supplementary irrigation at Konni, 1980-1997.



Yields of irrigated sorghum and millet have historically been 3X to 4X higher than rainfed yields in the Konni area of Niger. . .

Drip Irrigation and Soil Sampling for Plant Nutrients in Afghanistan

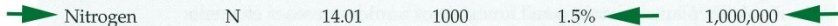


Nitrogen Deficiency

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Source: Modified from Stout 1961.



Calcium Deficiency

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Iron Deficiency

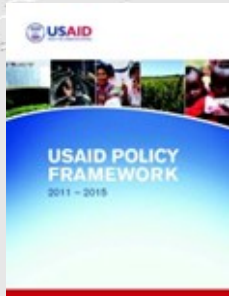
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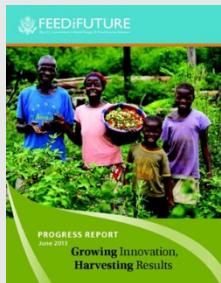
Source: Modified from Stout 1961.



USAID Promotes Scientific Innovation



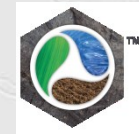
- “Science, technology, and innovation can produce particularly powerful outcomes when complemented by other investments.”



- “Scientific innovation and technology are critical to meeting the global challenges of producing more food with less **land** and **water** and helping farmers adapt to climatic, social and economic shocks.”



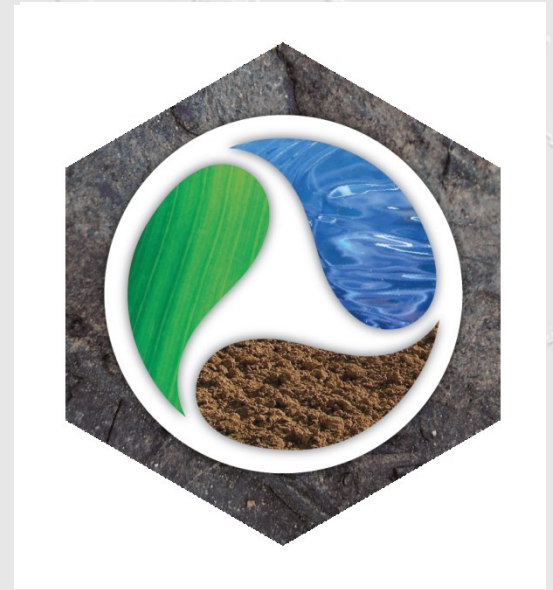
Micro Carbon Technology®



An Innovative Technology of Bio Huma Netics, Inc. (BHN) for
Fertilizers, Pesticides and Wastewater Treatment Products

Scientifically Engineered Technology

- Micro Carbon Technology®
 - Symbol
 - Carbon Ring (leonardite / carbon source)
 - Soil
 - Plants
 - Water



The Origins of Micro Carbon Technology®

- In 1973 scientists of Sunburst Mining Co., the predecessor of Bio Huma Netics, Inc., discovered a unique material in a company owned mine located in the state of Idaho.
- When applied to farmers' fields, this material improved both the soil's fertility and the plants' nutrient uptake.
- This material is mainly composed of leonardite, an oxygen-rich form of soft coal that is made up of decomposed plant matter and minerals.



Micro Carbon Technology®

An Innovative Fertilizer Technology

- Proprietary Extraction Process
 - A proprietary, time-tested biological & chemical extraction process.
 - Decomposes humic substances of leonardite.
 - Produces a mixture of organic molecules, many of which are small and more chemically active.



From Leonardite to Micro Carbon Technology®



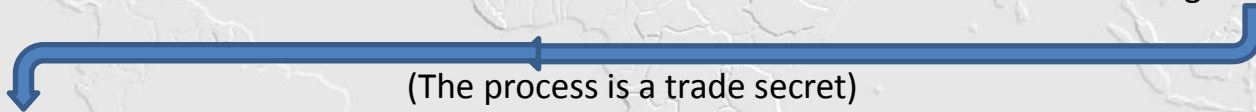
Leonardite Mine



Leonardite Delivered to BHN



Biological Digestion



(The process is a trade secret)



Chemical Extraction



Complexing with Nutrients



products



products

Functional Groups of Humic and Fulvic Acids that Contain Oxygen

	Total Cation Exchange Capacity	-COOH	Acid -OH	Weakly Acid and Alcoholic -OH	-C=O
	Normal Range, cmol(+) per kg				
Humic Acids	500-870	150-300	250-570	270-350	90-300
Fulvic Acids	900-1,400	610-910	270-670	330-490	110-310

Source: F. J. Stevenson and J. H. A. Butler. 1969. Chemistry of humic acids and related pigments. P. 534-577. In G. Englinton and Sister M. T. J. Murphy (eds.). Organic geochemistry. Springer-Verlag, Berlin.



Micro Carbon Technology® produces organic (carbon-containing) molecules more chemically active than humic and fulvic acids



The 17 Essential Plant Nutrients

PERIODIC TABLE OF THE ELEMENTS

1a	2a	3b	4b	5b	6b	7b	8	1b	2b	3a	4a	5a	6a	7a	0	Orbit		
1 H 1.0080 +1 -1															2 He 4.00260 0	K		
3 Li 6.94 2 1	4 Be 9.01218 2 2									5 B 10.81 2-3	6 C 12.011 2-4	7 N 14.0067 2-5	8 O 15.9994 2-6	9 F 18.99846 2-7	10 Ne 20.17 2-8	K L		
11 Na 22.98977 2-3 1	12 Mg 24.305 2 8 2									13 Al 26.9815 2-8-3	14 Si 28.086 2-8-4	15 P 30.9738 2-8-5	16 S 32.06 2-8-6	17 Cl 35.453 2-8-7	18 Ar 39.948 2-8-8	K-L-M		
19 K 39.10 8-8-1	20 Ca 40.08 8-8-2	21 Sc 44.959 8-9-2	22 Ti 47.90 8 10 2	23 V 50.94 8 11 2	24 Cr 51.996 8-13-1	25 Mn 54.938 8-13-2	26 Fe 55.847 8-14-2	27 Co 58.9332 8-15-2	28 Ni 58.71 8-16-2	29 Cu 63.546 8-18-1	30 Zn 65.38 8-18-2	31 Ga 69.72 8-18-3	32 Ge 72.59 8-18-4	33 As 74.9216 8-18-5	34 Se 78.96 8-18-6	35 Br 79.904 8-18-7	36 Kr 83.80 8-18-8	L-M-N
37 Rb 85.467 18-8-1	38 Sr 87.62 18-8-2	39 Y 88.905 18-9-2	40 Zr 91.22 18-10-2	41 Nb 92.9064 18-12-1	42 Mo 95.94 18 13-1	43 Tc 98.9062 18-13-2	44 Ru 101.07 18-15-1	45 Rh 102.9055 18-16-1	46 Pd 106.4 18-18-0	47 Ag 107.868 18-18-1	48 Cd 112.40 18-18-2	49 In 114.82 18-18-3	50 Sn 118.69 18-18-4	51 Sb 121.75 18-18-5	52 Te 127.60 18-18-6	53 I 126.9045 18-18-7	54 Xe 131.30 18-18-8	M-N-O
55 Cs 132.9055 18-8-1	56 Ba 137.34 18-8-2	57 La 138.9055 18-9-2	72 Hf 178.49 32-10-2	73 Ta 180.947 32-11-2	74 W 183.85 32-12-2	75 Re 186.2 32-13-2	76 Os 190.2 32-14-2	77 Ir 192.22 32-15-2	78 Pt 195.09 32-16-2	79 Au 196.9665 32-18-1	80 Hg 200.59 32-18-2	81 Tl 204.37 32-18-3	82 Pb 207.2 32-18-4	83 Bi 208.9806 32-18-5	84 Po (209) 32-18-6	85 At (210) 32-18-7	86 Rn (222) 32-18-8	N-O-P
87 Fr (223) 18-8-1	88 Ra (226) 18-8-2	89 Ac (227) 18-9-2	104	105														O-P-Q

KEY TO CHART

Atomic Number → 50
Symbol → Sn
Atomic Weight → 118.69
Oxidation States → +4
Electron Configuration → -18-18-4

Transition Elements

Group 8

Transition Elements

Metalic Essential Nutrients (Yellow background)

Non-Metallic Essential Nutrients (Red background)

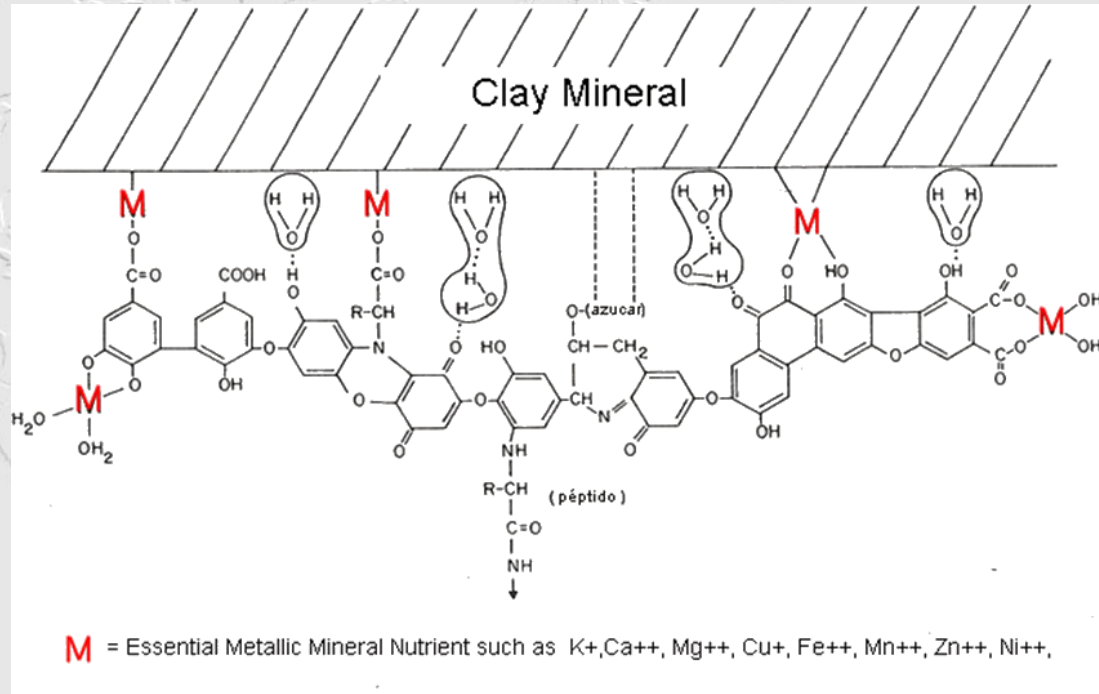
*Lanthanides	58 Ce 140.12 -20-8-2	59 Pr 140.9077 -21-8-2	60 Nd 144.24 -22-8-2	61 Pm (145) -23-8-2	62 Sm 150.4 -24-8-2	63 Eu 151.96 -25-8-2	64 Gd 157.25 -25-9-2	65 Tb 158.9254 -27-8-2	66 Dy 162.50 -28-8-2	67 Ho 164.9303 -29-8-2	68 Er 167.26 -30-8-2	69 Tm 168.9342 -31-8-2	70 Yb 173.04 -32-8-2	71 Lu 174.97 -32-9-2		N-O-P
**Actinides	90 Th 232.0381 18-10-2	91 Pa 231.0359 -20-9-2	92 U 238.029 -21-9-2	93 Np 237.0482 -22-9-2	94 Pu (244) -24-8-2	95 Am (243) -25-8-2	96 Cm (247) -25-9-2	97 Bk (247) -27-8-2	98 Cf (251) -28-8-2	99 Es (254) -29-8-2	100 Fm (257) -30-8-2	101 Md (256) -31-8-2	102 No (254) -32-8-2	103 Lr -32-9-2		O-P-Q

Numbers in parentheses are mass numbers of most stable isotope of that element.

Sources: Handbook of Chemistry and Physics, 54th ed. and Epstein and Bloom, 2005



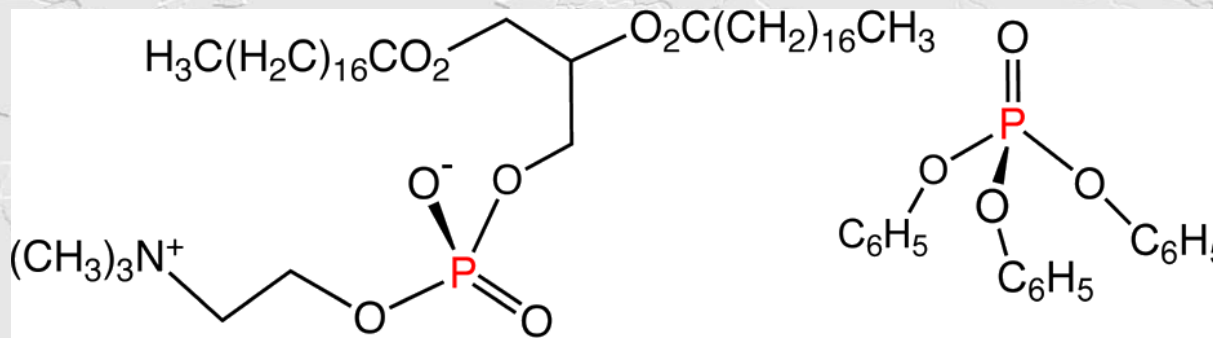
Clay-Metal-Organic Matter Complexes in Soil



Source: FJ Stevenson and MS Ardakani. 1972. Organic Matter Reactions Involving Micronutrients in Soils. In JJ Mordtvedt, PM Giordano and WL Lindsay (eds.) Micronutrients in Agriculture. Soil Science Society of America, Madison, Wisconsin.



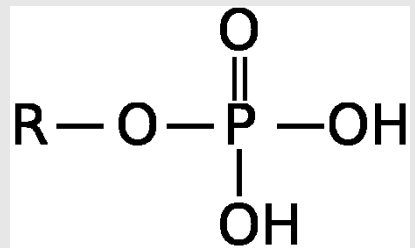
Examples of the Interaction of Phosphate with Organic (Carbon-containing) Substances



(Tertiary amine)

(Phenyl groups)

Phosphate esters*



*An ester is formed by condensation of an acid with an alcohol.

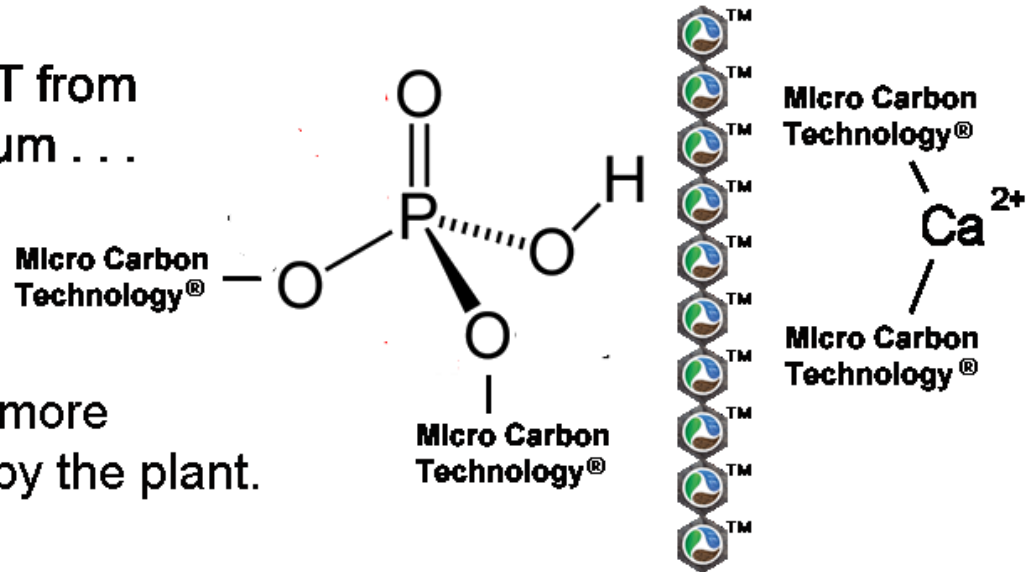


Increased availability of phosphorus from Super Phos™ in comparison with fertilizers of conventional phosphoric acid.

Phosphoric acid protected by Micro Carbon Technology® of Super Phos™

... is protected by MCT from precipitation with calcium ...

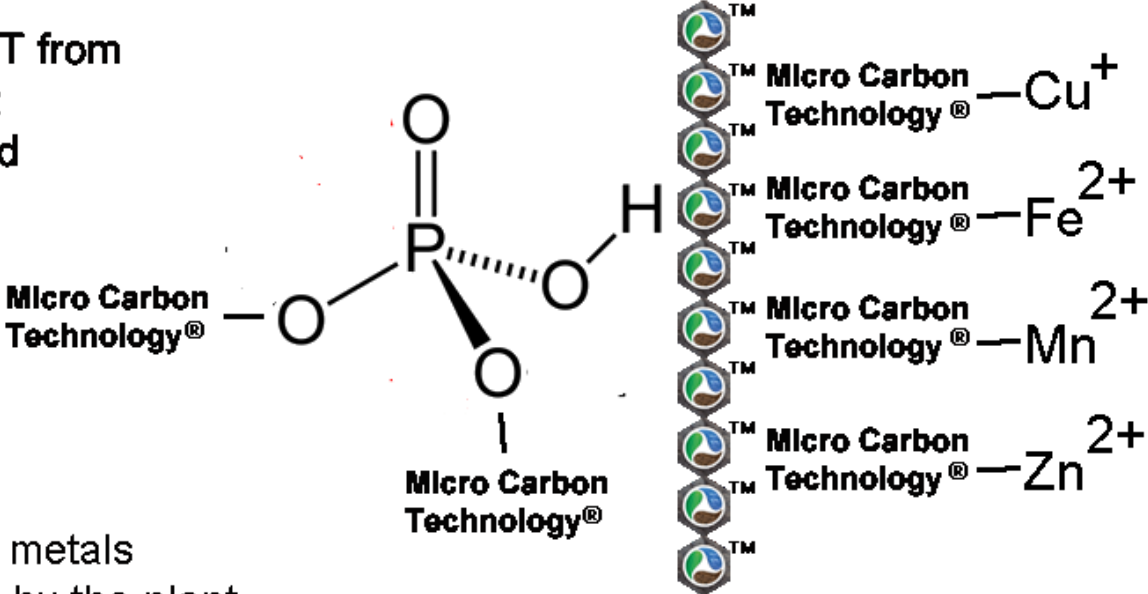
... making phosphorus more available for absorption by the plant.



Increased availability of micronutrient metals from Super Phos™

Micronutrient metals are protected by Micro Carbon Technology® of Super Phos™

Phosphate is protected by MCT from precipitation with micronutrient metals such as Cu, Fe, Mn and Zn . . .



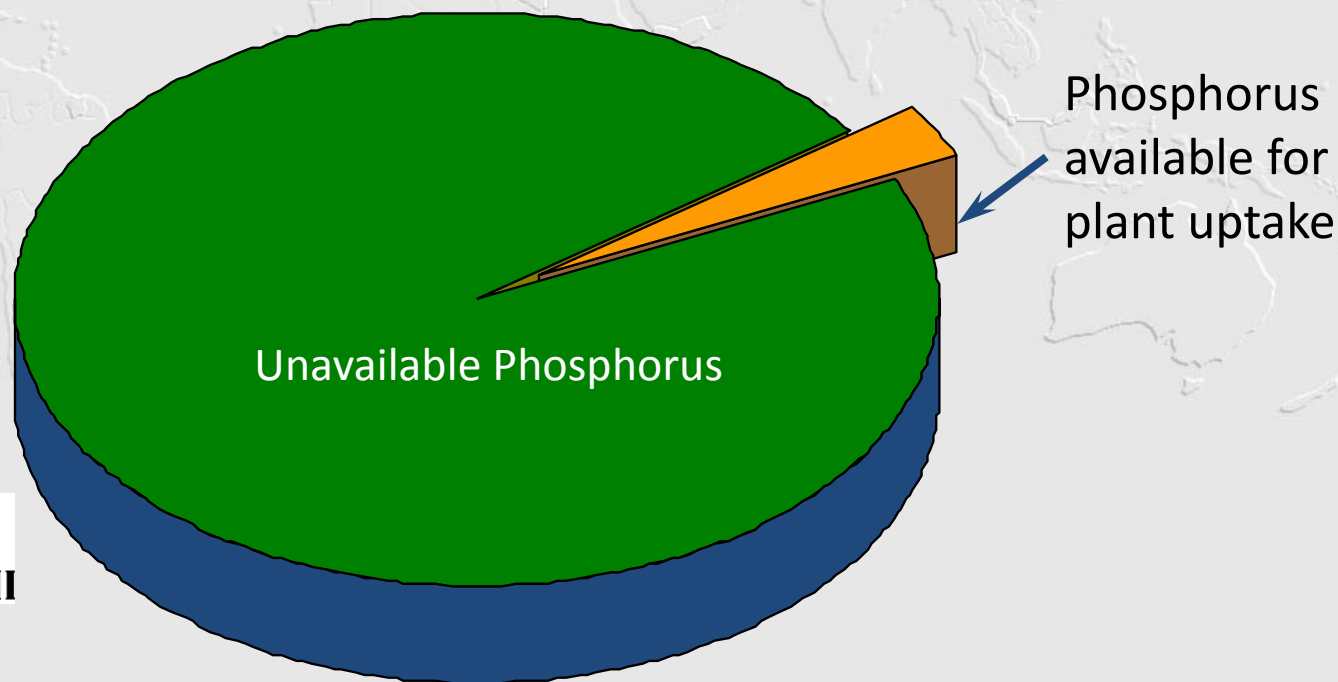
. . . making the micronutrient metals more available for absorption by the plant.



Phosphorus Inefficiency & Efficiency

Phosphorus is very insoluble in the soil, so only a fraction from most P fertilizers is available to the crop.

Total phosphorus in soils (0 to 6 inches) ranges between 400-2,000 pounds/acre . . . but only a fraction of that is available to the plant for the absorption each season.



Phosphorus Deficiency

TABLE 3.3 Adequate Concentrations of Elements in Plant Tissue

Element	Chemical symbol	Atomic weight	Concentration in dry matter		Relative number of atoms with respect to nickel
			$\mu\text{mol g}^{-1}$	ppm or %	
Micronutrients					
Nickel	Ni	58.69	0.001	0.05 ppm	1
Molybdenum	Mo	95.95	0.001	0.1 ppm	1
Cobalt	Co	58.94	0.002	0.1 ppm	2
Copper	Cu	63.54	0.10	6 ppm	100
Zinc	Zn	65.38	0.30	20 ppm	300
Sodium	Na	22.91	0.40	10 ppm	400
Manganese	Mn	54.94	1.0	50 ppm	1000
Boron	B	10.82	2.0	20 ppm	2000
Iron	Fe	55.85	2.0	100 ppm	2000
Chlorine	Cl	35.46	3.0	100 ppm	3000
Macronutrients					
Silicon	Si	28.09	30	0.1%	30,000
Sulfur	S	32.07	30	0.1%	30,000
Phosphorus	P	30.98	60	0.2%	60,000
Magnesium	Mg	24.32	80	0.2%	80,000
Calcium	Ca	40.08	125	0.5%	125,000
Potassium	K	39.10	250	1.0%	250,000
Nitrogen	N	14.01	1000	1.5%	1,000,000
Oxygen	O	16.00	30,000	45%	30,000,000
Carbon	C	12.01	40,000	45%	40,000,000
Hydrogen	H	1.01	60,000	6%	60,000,000

Source: Modified from Stout 1961.



Phosphorus Fertilizers Inefficient vs. Efficient

Addition of 10-34-0 to irrigation water resulted in formation of insoluble calcium phosphate



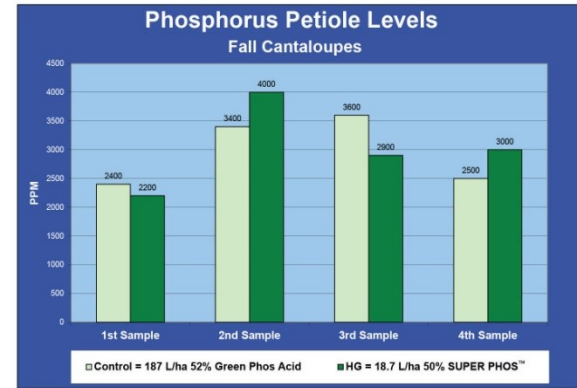
Calcium phosphate
white precipitate

10-34-0

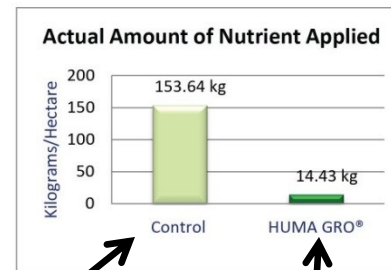
Clear water

FLOW

Inefficient Ammonium Polyphosphate
(APP, or 10-34-0)



HUMA GRO® SUPER PHOS™ product is so effective and efficient, it took 18.7 liters per hectare of SUPER PHOS™ (50% P₂O₅) to get the above results of phosphorus petiole levels in the plants. The SUPER PHOS™ amount used was only 10% of the control amount!




Inefficient phosphoric
acid fertilizer

Efficient Huma Gro
Super Phos™

Advantages of Micro Carbon Technology® (MCT) in Liquid Huma Gro® Fertilizers

- In the soil, MCT protects applied nutrients from loss due to formation of insoluble compounds.
 - Increases essential plant nutrient efficiency and water use efficiency.
- Liquid Huma Gro® fertilizers with MCT can be applied directly to the leaves,
 - Placing needed, essential nutrients very close to the principal site of photosynthesis.
 - Avoiding losses of nutrients in the soil.
 - Increases essential plant nutrient efficiency and water use efficiency.





Bio Huma Netics, Inc.
Global Value Chains Delivering Products
with Micro Carbon Technology®

**HUMA
GRO** products

 products

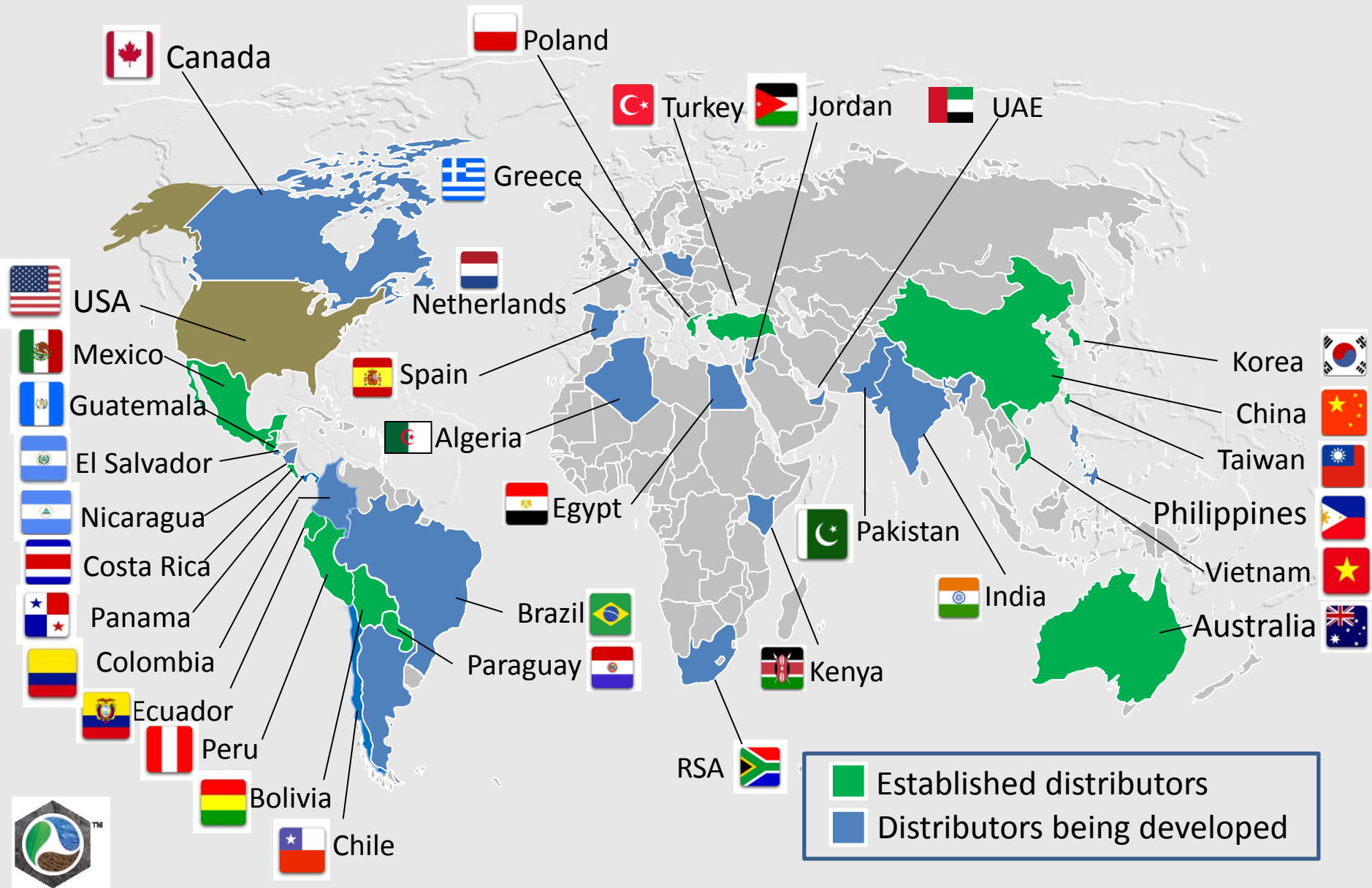


Manufacturing BHN Products

Fertilizers, pest control and wastewater treatment products



2013 Worldwide Presence of Bio Huma Netics, Inc.



Greatest Benefits (Yield and WUE) in Areas with Access to Soil, Water and Plant Testing



Steps to Opening a New Country

- Initial Contact
- Face to face meetings
 - Dist training, grower visits, infield work
- Send samples of key products
- Begin registration process (if necessary)
- Begin field demonstrations of products on key crops
- Products Registered
- First container ordered
- Commercial



Value Chains – Markup to include costs and profit

- Manufacturer suggests retail cost of 100 units of local currency where product is to be used/unit volume of product (MSRP).
- Master distributor buys from manufacturer at 40 units of local currency/unit volume delivered to distributor's port.
- Distributor pays between 10 units and 25 units of MSRP to import (duties, taxes).
- Distributor sells to sub-distributor for 65 units to 70 units of MSRP.
- Sub-distributor sells to retailer for 70 units to 75 units of MSRP.





BHN Value Chains in Spain



Regional Huma Gro[®] Supply Chains in Spain

Manufacturer

Crucial business relationship
with master distributor

Exclusive, performance-based
contracts with regional distributors

In each region, the regional
distributor has business relationships
with local distributors, commissioned
agents, and/or end users



HUMA GRO[®] HISPANIA S.L.



CUALIN

FRABELSE

GAMBAO

SALQUISA

VALSECO

COMPANY





BHN Value Chains in China



Packaging BHN Products in Sizes Appropriate for Small-Scale Farmers in China



Application of Huma Gro® Fertilizers to Kiwifruit in China





BHN Value Chains in Nicaragua



Application of Huma Gro[®] Fertilizer to Coffee Seedlings in Nicaragua



Close-Up View of Application of Huma Gro® Fertilizer to Coffee Seedlings in Nicaragua





**Wastewater Reuse from Small-Town Sewage, Dairy, and Abattoir
Enhanced Treatment for Crop Irrigation**

Economic Benefits of Bio Energizer[®]-Reclaimed Small-Town Sewage in Texas

- **Raw sewage** (“municipal wastewater”) passes through a bar screen at the headworks to capture rags, bags, wood and other large objects.
- The sewage proceeds through a series of 3 lagoons to decrease solids and increase available plant nutrients.
- After the water has been in the system for approximately 30 days, it is pumped from the last lagoon and applied to a hay field via sprinklers on a typical irrigation schedule.
- *The treated water has been an economic boon to the farmer who bails the hay and sells it to the ranchers who no longer have to purchase hay produced elsewhere.*

Simple bar screen



Headworks
and bar screen



Lagoon treated with Bio Energizer[®]

Irrigation pivot in background



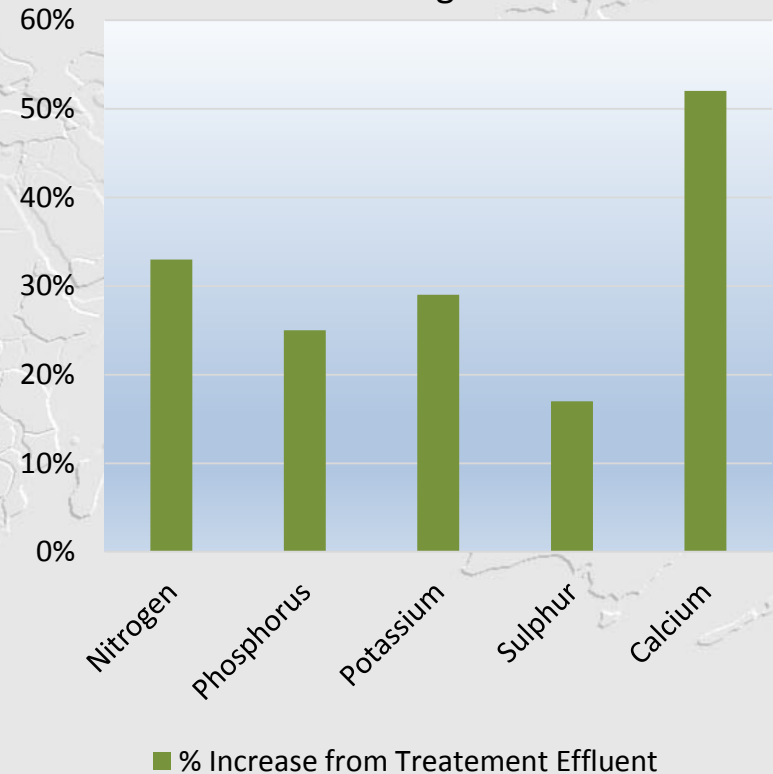
Irrigation of forage with
treated wastewater

Improvement of Forage Quality Using Dairy Wastewater Treated by Bio Energizer® in Australia

Irrigated *Phalaris* pasture grass



Increase in Plant Tissue Concentrations Resulting from Effluent Treated with Bio Energizer®



Qualitative and quantitative improvements in *Phalaris* pasture grasses irrigated with effluent pond water treated with Bio Energizer®, compared to a control pasture irrigated with non-treated **dairy wastewater**.

Treated Wastewater from Ovine Abattoir when Treated with Bio Energizer® Improves Wheat & Canola Production in Australia

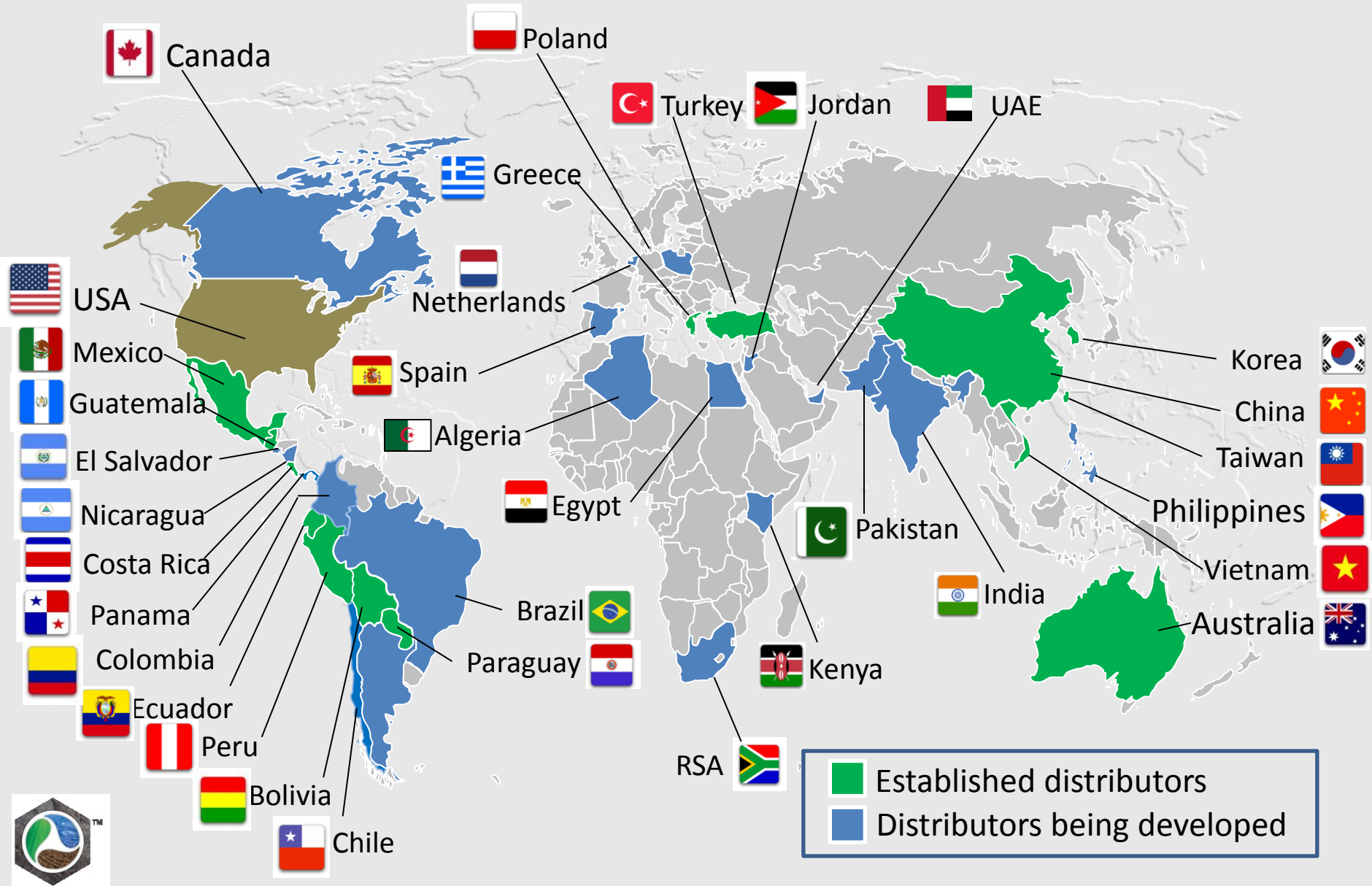


Treated wastewater from a **ovine abattoir** is drawn off the last “polishing” treatment pond below and pumped to an adjacent field where crops are rotated annually. After using Bio Energizer® in the pond system for a year the abattoir won “Wheat Crop of the Year”. And this, year 2013, the operator exclaimed “You should see my canola crop it is this high” – gesturing with his hand to about 3.5 feet high.

The operator of the abattoir’s wastewater treatment facility was clearly excited about the growth of his latest crop, due to the addition of Bio Energizer® into the treatment pond system as compared to previous years without it.



2013 Worldwide Presence of Bio Huma Netics, Inc.





Thank you for your interest.

Additional information: <http://bhn.us> or tom@bhn.us

